



Sai College®

COURSE OUTCOME S
OF
BACHELOR OF SCIENCE

VISION

Our vision is to produce competent Biotechnologists who can employ premium processes and applications which will profoundly influence existing paradigm of agriculture, industry, healthcare and restoration of environment providing sustainable competitive edge to present society.

MISSION

- To provide Biotechnology educational Program with impetus to generate quality workforce.
- To create awareness about potentials of Biotechnology with socio-ethical implications.
- To instill spirit of innovation and creativity in young minds with sound research aptitude.
- To nurture confident individuals who are effective contributors towards growth of the nation.

SAI COLLEGE**Program - B.Sc. (CHEMISTRY)****PROGRAM OUTCOMES**

PO-1. Understand the basic concepts of science and able to correlate them with their daily life.

PO-2. Develop skill of handling instruments, performing experiments and logical analysis of the experimental results.

PO-3. Analyse the every situation of day to day life with scientific approach and able to draw objective conclusion for the betterment of society and humanity.

PO-4: To educate students on topics: Periodic Classification of Elements and Chemical Bonding

PO-5: To educate students on basic organic chemistry of saturated and unsaturated hydrocarbons along with introduction to different types of reactions.

PO-6: To expose students on radical analysis in inorganic mixtures; determination of surface tension and viscosity of liquids.

PO-7: To make students aware about 's', 'p' and Zero Group elements and their related chemistry

PO-8: To educate students on the topics: chemical kinetics, photochemistry and macromolecules.

PO-9: To provide laboratory experience to the students by performing volumetric analysis for determination of equivalent weight of acids and experiments based on organic compounds synthesis and their purification methods.

PO-10: To provide opportunity and experience of presenting seminar on pre-allotted topics related to theory.

B.Sc. (Chemistry)

Paper	Name of Paper
Paper I	Inorganic Chemistry
Paper II	Organic Chemistry
Paper III	Physical Chemistry

Course Outcomes

Paper	Paper Name (Paper code)	Outcomes
B.Sc. I		
Paper I	Inorganic Chemistry	<p>CO –1 To get the knowledge of the structure of atom and periodicity in the properties of elements</p> <p>CO – 2. To get the knowledge of the process of formation of ionic bond and properties of ionic solids</p> <p>CO – 3To get the knowledge of the theories of covalent bond formation</p>

		<p>CO – 4. Know the salient features of s and p block elements</p> <p>CO – 5. To get the knowledge of the chemistry of noble gases and theoretical principles involved in qualitative analysis</p>
Paper II	Organic Chemistry	<p>CO – 1. To get the knowledge of the basic concepts and electronic effects of organic chemistry.</p> <p>CO – 2. Know the stereochemistry of organic molecules</p> <p>CO – 3. To get the knowledge of the conformational analysis of alkanes</p> <p>CO – 4. Understand the chemistry of alkane, alkene and alkyne</p> <p>CO – 5. Know the concept of aromaticity and electrophilic substitution reaction in aromatic compounds</p>
Paper III	Physical Chemistry	<p>CO – 1. To get the knowledge of the basic mathematical concept used in chemistry</p> <p>CO – 2. Understand the kinetic molecular model of gas and understand the behaviour of real gases</p> <p>CO – 3. To get the Knowledge of the intermolecular forces and understand colloid and surface chemistry</p> <p>CO – 4. Understand the symmetry, crystal system and crystal defects</p> <p>CO – 5. Know the rate of reaction, factors affecting it and theories of reaction rate and catalysis.</p>
Lab course	Inorganic chemistry	<p>CO – 1. To get knowledge of the inorganic mixtures by the Semi-micro qualitative analysis</p> <p>CO – 2. Understand the strength of unknown solution by volumetric method</p> <p>CO – 3. Get the knowledge of elements (N, S and halogens) and functional groups in organic</p>

		<p>compounds</p> <p>CO – 4. Understand the measurement and composition of a binary liquid mixture by surface tension method</p> <p>CO – 5 To know the composition of a binary liquid mixture by viscometer</p>
B.Sc. II		
Paper I	Inorganic Chemistry	<p>CO – 1. To get the knowledge of chemistry of transition series elements</p> <p>CO – 2. To get the knowledge of the redox potential data & its application and chemistry of coordination compounds</p> <p>CO – 3. Understand the valance bond theory and crystal field theory</p> <p>CO – 4. To know the chemistry of lanthanides and actinides</p> <p>CO – 5. Understand the theories of acid and bases and physical properties & chemical reactions of non-aqueous solvents</p>
Paper II	Organic Chemistry	<p>CO – 1. To get the knowledge of the mechanism of nucleophilic substitution and elimination reactions</p> <p>CO – 2. Understand the preparation, properties and reactivity of alcohol and phenol</p> <p>CO – 3. To Know the nomenclature, structure and reactivity of carbonyl group</p> <p>CO – 4. To get the knowledge of chemistry of carboxylic acid and its derivatives</p> <p>CO – 5. To get the knowledge of the reactivity, structure and properties of organic compounds of nitrogen</p>
Paper III	Physical Chemistry	<p>CO – 1. To know the laws of thermodynamics and know the meaning of various thermodynamic terms</p>

		<p>CO – 2. To get knowledge of the concept of entropy and free energy</p> <p>CO–3. Understand the chemical & ionic equilibrium and equilibrium constant</p> <p>CO – 4. To get the knowledge of the phase rule and its application to one, two and three component system</p> <p>CO–5. Get the knowledge of characteristics of electromagnetic radiation, laws of photochemistry and quantum yield</p>
Lab course	Chemistry Practical	<p>CO – 1. Understand the semimicro analysis of mixtures containing interfering radicals.</p> <p>CO – 2. Determine the strength of solution by volumetric method</p> <p>CO – 3. To get the knowledge of specific organic compound</p> <p>CO – 4. Understand the R_f value and identify organic compound through paper chromatography</p> <p>CO – 5. To get the knowledge the enthalpy of chemical reactions</p>
B.Sc. III		
Paper I	Inorganic Chemistry	<p>CO – 1. Know the metal-ligand bonding in transition metal complexes</p> <p>CO – 2. To get the knowledge of magnetic properties of transition metal complexes</p> <p>CO – 3. To Know the classification, properties, bonding and applications of organometallic compounds</p> <p>CO – 4. Understand the essential and trace elements in biological processes</p> <p>CO – 5. To get the knowledge of concept of hard and soft acid and base and inorganic polymers</p>
Paper II	Organic Chemistry	<p>CO – 1. Understand the heterocyclic organic compounds</p>

		<p>CO – 2. To get the knowledge of organometallic compounds</p> <p>CO- 3 To get the knowledge of understand the knowledge</p> <p>CO – 4 Understand the chemistry of synthetic polymers and dyes</p> <p>CO – 5. Understand the principle and applications of NMR, IR and UV – Visible spectra</p>
Paper III	Physical Chemistry	<p>CO – 1. To get the knowledge of basic concept of quantum mechanics along with Schrodinger's equation & its applications</p> <p>CO – 2. Know the quantum mechanical approach of molecular orbit theory</p> <p>CO – 3. Get the knowledge of principle and applications of Microwave, Infrared and Raman spectra</p> <p>CO – 4. Understand the concept electrochemistry for electrolytes</p> <p>CO – 5. Understand the different types of electrochemical cell and their potential</p>
Lab course		<p>CO – 1. Synthesis of inorganic complexes</p> <p>CO – 2. Gravimetric estimation of element</p> <p>CO – 3. Synthesis of Organic Compounds</p> <p>CO – 4. Analysis of an organic mixture containing two solid components</p> <p>CO – 5. Determine the strength of acid or base by conductometric titration</p>